

Advancing Arson Investigation: A Comprehensive Review of Crime Scene Techniques, Analytical Methods, and Evidence Management

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Abstract: Catching arson has numerous constituents and is all about effectively establishing the scene, the cause, and how it occurred. This all-encompassing look highlights the evolution of crime scene tactics, analytical instruments, and management practices involved in managing arson investigations. The approach increasingly presents different technologies that can be utilized for fire scene documentation, including digital forensics and fire pattern analysis, as well as the significance of forensic chemistry in identifying accelerants. It also discusses evidence collection, preservation, and chain of custody methods considered best practices for maintaining the integrity of findings for use in litigation. The paper highlights how fire science and law enforcement skills are exploited with new analytical tools to make arson investigation an interdisciplinary field in precision and efficiency. It demonstrates how, through this analysis, ongoing development in the field may be noted and further recommendations made for improvement of investigatory outcomes concerning arson cases.

Keywords: Arson Investigation, Crime Scene Techniques, Accelerants, Analytical methods

I. Introduction

Arson is among the most complex and dangerous crimes to commit, posing great challenges to public safety, property, and the environment. It usually involves intentional firesetting with intent to cause damage; its investigation requires a rich understanding of fire behavior, forensic techniques, and legal considerations. Complexity in arson cases arises from certain inherent difficulties in scenarios of fire scenes where destruction often destroys critical evidence, hence demanding timely and accurate analysis.

It has become vital to advance arson investigation techniques so that perpetrators are identified and the cases are solved efficiently. Significant innovations in the field over the years include innovations in crime scene methodologies, analytical tools, and evidence management practices. The approach of arson investigation has changed by the application of scientific principles in the recognition of fire patterns, the use of accelerant detection, and advanced forensic technologies. In addition to this, great care must be taken in collecting and recording all evidence, and this is mainly because the court relies on its integrity to prosecute successfully.

It is in relation to this fact that this critical review seeks to discuss the sophisticated techniques and methodology applied in current arson investigations: crime scene analysis, analytical method, and control of evidence. This paper seeks to provide an overview of the state of arson investigation by synthesizing existing literature and discussing advancement in the field. The main goal is to identify potential areas for further research and improvement, which will better equip investigators to combat arson with the tools and knowledge required to understand these evolving practices.

Crime Scene Investigation

Crime scene protocols:

In the arson crime scene, the procedure adopted is followed in detail by the investigators to ensure that all evidence is collected and preserved and to reconstruct the origin and progression of the fire. The response is made of securing the scene, damage assessment, and preliminary survey to recognise potential fire patterns and other evidence. This scene investigators record in a mix of observations from the visible sense, photos, and video captures of the patterns of fires and patterns of burns in the burnouts or structural/fixture damages. The origin, spread, and development of fire in the entire scene shall be defined by analyzing the fire dynamics and behaviour analysis. These include the nature of fuels involved, origin, kind of ignition, and existence of accelerants. Techniques normally employed include utilizing canines, accelerant detection devices, and lab testing of debris and other evidences. (Franjic, S.2018)

Evidence is collected and preserved at the crime scene where arson may take place to reconstruct the origin and spread of the fire. There is an initial response followed by securing the crime scene to prevent contamination. Assessment of the damage done during the process of reconstructing the origin and spread is made, followed by going through every minute detail starting with photographic records of visual observations concerning patterns of fire and damages incurred. (Lentini, J.J.2012)

Understanding fire dynamics is important because investigators use burn patterns to tell where the fire came from, what fuel it burned, and how many potential ignition sources are at that location. Investigators also often detect accelerants with canines and through lab work on debris. The evidence collection process entails meticulous samples of burned materials, which forms an integral part of how the various materials burn in an accelerant. (Chi, J. H., & Peng, P. C. 2016)

Crime scene investigation is a very technical process requiring multidisciplinary expertise and has to be detailed and procedure-based. Ensuring that the evidences are not damaged before their preservation should lead the procedure in any investigating process.

This can be achieved through the formation of a restricted perimeter whereby barriers or tape are utilized to bar unauthorized persons from entering the scene. Even the emergency responders, who might have unwittingly changed the scene in their process of arriving, are photographed so that there exists an undisturbed record of possible alterations. The first walk-through is generally referred to as a preliminary survey. This provides the investigator with a general view of the scene. Here, fragile or ephemeral evidence, like blood spatter, footprints, or accelerant residues, are photographed. These are critical clues in building the initial hypothesis concerning the sequence of events leading to the crime. (Cheenmatchaya, A., & Kungwankunakorn, S. 2018)

Documentation starts methodically after securing the scene using a combination of high-resolution photography, sketching, and written descriptions. Photographs would have details of burn patterns, soot, or the dislocation of objects while a sketch would give out the spatial relationships and dimensions. Written notes record observations, environment conditions, and any interpretation that the investigator may personally make. All these ensure a strong record that might be referred to during laboratory analysis or in court. Evidence is collected in a very elaborate process to preserve integrity so that cross-contamination should not occur. For instance, in arson cases, debris samples are taken from the suspected points of origin using sterilized equipment, as well as samples of accelerants and fire residues, which are packed in sealed containers. The control samples are usually taken in the form of unburned material from the same source to be used as a comparison during analysis in the laboratory. (Chi, J.H 2013)

Crime scene investigation involves collaboration. It would take all fire investigators, forensic chemists, and accelerant detection specialists to make sure that all evidence is evaluated. Detection dogs that are trained to detect ignitable liquids are critical as they tend to be much more sensitive and accurate compared to the detection equipment. Portable gas chromatographs and spectrometers are also brought to the scene of action to do in-field chemical analysis supporting the decision to investigate further. All data, observations, and interpretations shall be summarized in detail to become the bedrock for presenting legal arguments and delivering expert testimonies. This report is documented through photographs, analytical results, sketches, and procedural details to make the whole exercise open and defensible. (Muller, D., Levy, A., & Shelef, R. 2014)

Fire dynamics and fire behavior:

Knowledge of fire dynamics is, therefore, very important in any line of inquiry regarding fire incidents as well as differentiating between accidental fires and intentional ones. It describes fires by three essentials: fuel, oxygen, and heat. Fuel comprises any combustible material - wood, fabric, and synthetic polymers, but the source of oxygen primarily comes from air. Then again, the inflow of air and ventilation may determine the strength of the fire. The ignition source is some kind of heat that brings chemical action for the combustion process. It just happens that if any of these is removed the fire gets extinguished, and that's why this phenomenon of fire fighting occurs. (Muller, D., Levy, A., & Shelef, R. 2011).

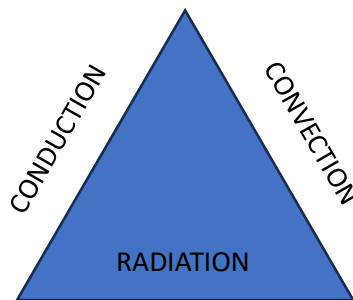


Fig 3.1: Fire triangle

Fire generally spreads with the aid of three phenomena conduction, convection, and radiation. Heat passes by conduction through solid objects like a metallic beam or wall. Convection occurs at the place where hot air ascends. This takes the flames and hot air into higher planes or other rooms. Radiation involves transferring heat energy to other objects which can also ignite without touching the fire. Experts analyze the mechanisms to get an insight into how a fire had spread in a building or location. It is thus essential in understanding the cause of a fire and even more so the behaviors. (Bian, C. 2018)

The life cycle of a fire gives more insight into its characteristics. In the incipient stage, this fire commences from a small source of ignition and gradually evolves to consume oxygen and fuel. In the growth phase, the fire develops very rapidly in terms of rate development depending on the levels of ventilation and combustible material available. In the developed stage, it refers to the peak intensity of the fire, which is usually at maximum heat output and flame height. Lastly, in the decay stage, it is at the point flames begin to consume their fuel; however, there is a small probability of re-ignition by smoldering embers. (Kebakaran, T. 2018)

In general, fire works in such a way that it produces unmistakable physical evidence. This physical evidence is used to form the cornerstone of reconstructive investigation. Burn patterns, for example, can indicate the point of origin of the fire. A V up against the wall suggests a strong source of heat, and irregular shapes in burns could indicate that some accelerant has been used. Another highly useful clue is the characteristic properties of smoke: dark, heavy smoke tends to indicate involvement of synthetic materials, plastics. Light-colored smoke indicates source from natural combustibles, wood or paper. The size and direction of the thermal degradation also serve to indicate a possible pattern and role in fire spreading of wind and ventilation. (Choi, S., & Yoh, J. J. 2017)

Advanced forensic techniques involve high-technology analyses of the evidence of fire. Accelerant residues can be detected, even at trace levels using a general GC-MS procedure. Infrared and Raman spectroscopy offer non-destructive analytical methods of fire residues without damage to the evidence. LIBS is one technology that allows for the real-time analysis of residues from fire debris and combustion.

It enables investigators to detect chemical composition of residues, so they could ascertain whether it had accelerants present hence enable them to conclude whether it is an accident or a premeditated action. (Maurer, M. K., Bukowski, M. R., Menachery, M. D., & Zatorsky, A. R. 2010)

Understanding the dynamics behind fires also helps people find the psychology and behavior linked with arson. The point of origin, the accelerant used, and even the unique method of ignition often mark a pattern that is then capable of connecting the perpetrator with other incidents. Serial arsonists, for example, leave some sort of identifiable signature often including the utilization of some kind of accelerator, along with the selection of locations they burn. The integration of fire dynamics with behavioral analysis benefits as an investigator is able to identify an offender and trace a possible perpetrator. (Touroo, R., & Fitch, A. 2018).

Evidence Collection

Physical evidence on arsonists can include ignition implements, burned patterns, and residual accelerants on clothing, shoes, or skin. Hands can be soiled when a flammable liquid is deliberately poured on the ground or furniture. The study involved collecting gloves or filter paper, placing them in Nylon bags, and conducting a passive headspace extraction and concentration of volatile compounds in an oven at 60 degree Celsius for 16 hours. The DFLEXTM device, consisting of activated charcoal strips, was used to standardize adsorption conditions. The elution was performed with a 0.5% mixture of 2-hexanone and carbon disulfide(Melinda Darrer,2008).

Samples collected from various structures, including plastic bottle remnants, One Pot solid waste, wood, carpet, and pad sections likely containing solvent. The solid material was removed and placed in a glass vial, while wood samples were cut and placed in metal cans. Wall wipe samples were collected for drug residue. Burned and unburned exemplars were prepared and stored separately.(Matthew K. Green,2017).

Collection of entomological evidence

Fires can't destroy forensic evidence, especially entomological evidence, which can be used to estimate elapsed time since death after an arson, as disposing of a body is challenging.(G.S. Anderson,2005). For capturing the adult insects, use an insect net and transfer into a jar for preservation. Larvae samples are collected from the remains and preserved them in alcohol.

A study on the persistence of unleaded petrol on carpet ,found that small volumes are unlikely to be detected after 24 hours, while larger volumes are generally undetectable after 1 week. A known history study found that petrol is not typically found on previously uncontaminated carpet mats after a 6-week period, but the occupants occupation and behavior can affect the compounds deposited. An unknown history study found that only a small proportion of motor vehicles exhibit petrol on carpet or mats(K. Cavanagh,2002)

This study demonstrates that forensically valuable entomological evidence can be preserved even in a crime-intentional arson. Entomological evidence can indicate the victim's death, estimate the minimum PMI, and estimate the fire's time. However, caution is needed when considering later successively colonizing species, as the fire may disrupt their normal colonization patterns. Entomological remains survived on skeletonized and decayed car trunks, and even dispersed remains in the passenger compartment. Fire walls in vehicles reduced access for insects and limited fire damage(Stacey L. Malainey,2019).

Entomological evidence can persist through a fire, but the surviving insect remains are fragile. If a fire is deliberately extinguished, the insects will be wet and degrade rapidly. A forensic entomologist is needed to collect evidence. If collected shortly after the fire is extinguished, the water and heat kill larvae, making it less damaging. Pupae inside intact and empty puparia become brittle, so they must be handled by an experienced collector, preferably a forensic entomologist(Gail S. Anderson,2020).

Instrumental techniques are not sensitive enough to gather all information in samples, and current testing methods may lose valuable information due to different ignitable liquids chemical profiles and the complex nature of fuels. Detailed chemical fingerprints of fresh and aged petroleum oils have great potential in forensic investigations. Gas chromatography coupled with flame ionization detection (GC-FID) or mass spectrometry (GC-MS) is currently used to analyze samples in arson cases(Jessica Pandohee,2020).

Pre-search Planning and Preparation

- Assess the fire location.
- Gather relevant information form a search team.
- Plan detailed preparation based on the case's circumstances.

Equipments required for the investigation includes Digital camera (minimum 16 mp), Personal protective equipment, Flashlight and spot light, Writing materials, Multipurpose tools (screwdrivers, wire cutters, knives), Rubber gloves, Shovel or other hand tools, AC voltage tester, Tape recorder/voice recorder, Portable equipment for detection of hydrocarbon vapour buried under debris.

Evidence collection kit includes Barrier Tape, Forceps, Gloves and cotton gauge, Paper envelopes (6×6 cm., 15×10 cm.), Airtight metal (tin) containers of various sizes and cardboard box, Magnifying glass, Scissors, screw driver, wire cutter, Evidence tag, Mark in cloth, needle and thread, Match box, candle and sealing wax bar, Marking pen, Standard metallic seal of IO.

Footprint kit – includes Footprint lifters, Electrostatic footprint lifting device, Plaster of Paris, Machine oil, Talcum powder, Spirit, Shellac solution, Wire mesh, Frame for preparing plaster cast, Magnetic compass, Glass sheet (35×20 cm.), Transparency sheet/tracing paper, Fine tip permanent marker, NIC.

Fingerprint kit includes Fingerprint powders (black, grey, anthracene, magnetic, fluorescent), Brush (camel hair, ostrich hair, magnetic), Light source, Adhesive tape, Fingerprint lifting cards, Gloves.

Protection of the crime scene

The crime scene must be safeguarded from unauthorized entry to prevent alteration, destruction, loss, or contamination of evidence and prevent injury due to the building's fire or weak structure. ● Safeguard the location as quickly and as effectively as possible.

● If injured/burnt body shows any sign of life, provide immediate medical help and transport the body to the hospital, taking caution so as to disturb the scene to the minimum extent.

- All unauthorized persons (relatives, friends, onlookers, photographers, etc.) should be excluded from the scene.
- Officer-in-charge of the case should determine when, why and who should enter the scene.
- Police officer should not leave the scene unguarded till scientific examination is completed.
- The witnesses, suspects and informant should be detained for further details.
- Crime scene should not be altered; it should be barricaded, enclosing a larger area than smaller one so that no clues are left out.
- Officer should not introduce any material, like cigarette ends, foot or footwear marks, fingerprints, etc.
- During examination the original condition of doors, windows, stair-case, lighting, routes of entry/exit should be preserved.
- Do not move anything from its place. It has to be described and located by sketches and photographs.
- Scene should not be cleaned till thoroughly examined.

Photography of the Crime Scene

Photographs of the following should be taken

- The entire fire scene from different angles.
- Places where fire caused damages.
- Place where the maximum damage has taken place.
- Smoke on walls, roof and burnt patterns
- Suspected electrical short circuit in wires, fuse box, switches, etc.
- Burnt wires.
- Possible route of approach of criminal.
- Points of entry and exit.
- The location of buried/dead body.
- Evidence left by criminal at the scene (cigarette buds, fingerprints, foot/footwear marks, etc.)
- If possible, videography at the scene should be done.
- Presence of any foreign matter (bottle/container, matchstick, cigarette/bidi, burnt rope, etc.)
- Burnt electrical appliance and surrounding.
- Close up photographs to cover site of fire and to cover important points and evidences.

Sketching of the Crime Scene

Sketches are to be drawn by an expert/IO and following are to be included in the sketch.

- The entire area where fire was put up.
- Inter-distances between relevant objects and evidences.

- Location of seat of fire.
- Place, date and time when the sketch was made.
- Name of persons assisting in taking measurements.
- Directions (east, west, north and south)

Packaging of evidence

1. The evidence should be packed in the container having qualities as below
 - Container should be made of metal like unused paint cans which is made of tin.
 - Air tight container should be used to avoid evaporation.
 - Container should be resistance to breakage.
 - Good integrity seal.
2. Collection of smaller samples can be done with glass container with metal cap.
3. Paper bags, polythene envelope, plastic container should not be used for packaging.
4. Collect more then two quarts of ash, soot and debris from origin of fire .
5. Leaving space on the top of the container by filling up only half or two third of the evidence in a container.
6. Samples like wood, furniture, cloth etc., must be packed in air tight container with clean labeling.
7. Liquid samples or fuels which are found in open bottles can be placed in glass container to avoid loss of fuel.
8. Burnt wires should be placed in thermal sheets and packed in plastic box or cardboard box.
9. Electrical appliances can be packed in cardboard boxes .
10. Cast of footprints or shoeprints must be packed and labeled properly.

Analytical Techniques for Detection of Arson Residues

RESIDUE TYPE	ANALYTICAL TECHNIQUE	COMMON USES
Organic Residues	GC-MS, FTIR, Raman Spectroscopy, HPLC, SPME, SERS	Identifying accelerants (e.g., gasoline, alcohols, solvents)
Metallic Residues	ICP-MS, EDS, XRF, AAS, NAA	Detecting metals in incendiary devices, explosives, or metal-based accelerants
Trace Residues	GC-MS, LA-ICP-MS, SERS, SEM-EDS, TEM, Ion Chromatography, XRD	Detecting minute amounts of organic, metallic, or ionic residues left behind in debris

Table 5.1: Categorization of arson residue types and their analytical techniques.

Organic Residues

The examination of combustion byproducts is crucial in fire investigations, as it yields significant insights into the materials involved and the possible use of accelerants. Forensic chemists utilize analytical chemistry techniques to extract, isolate, and analyze specific compounds that define ignitable liquid residues or the nature of fuels. A critical consideration in forensic examinations is the sensitivity of the techniques employed, which determines the minimum concentration of target analytes necessary for definitive identification. Forensic chemists engage in analytical chemistry to extract, isolate, and scrutinize the compounds that signify ignitable liquid residues or the characteristics of fuels.

In forensic fire residue analysis, the detection of ignitable liquids is of utmost importance, thus making sensitivity a critical consideration. Forensic chemists apply principles of analytical chemistry to extract, isolate, and analyze the compounds that are indicative of ignitable liquid residues or fuel types. In the context of forensic fire residue analysis, the identification of ignitable liquids is particularly critical, making sensitivity a significant concern. (Niculina-Sonia Suva, 2023)

ATR-FTIR

The infrared beam is reflected off the undersurface of a refracting crystal, which then interacts with a sample of lower refractive index placed onto the upper side of the same surface. There is a tiny interaction between the sample material and the beam of light, due to the phenomenon known as an ‘evanescent wave’, which is shown in the diagram below. In order to ensure that the beam of

light is reflected rather than transmitted through the surface of the ATR crystal the angle of incidence must be above the critical angle, which varies depending on the refractive index of the crystal material. In traditional ATR-FTIR spectroscopy, a high-refractive-index crystal – such as diamond, germanium, or zinc selenide – is used as the ATR element.

This FTIR research carried out by Farah Izza Jais et al., indicates that the study highlights the forensic importance of using non-destructive Attenuated Total Reflectance-Fourier Transform Infrared (ATR-FTIR) spectroscopy combined with chemometric methods to distinguish different accelerants on various fabric types post-combustion or arson. Six fabric types (cotton, wool, silk, rayon, satin, and polyester) were burned using RON95 and RON97 gasoline, as well as diesel fuel. The charred samples were analysed with ATR-FTIR spectroscopy, and the data were further processed using Principal Component Analysis (PCA) and Linear Discriminant Analysis (LDA) to differentiate the accelerants present on the burned fabrics (Farah Izza Jais&Sharifah Mastura,2020).

GC-MS

Gas chromatography-mass spectrometry (GC-MS) is a combined analytical technique that integrates two procedures in sequence: Gas Chromatography (GC) for separating volatile compounds and Mass Spectrometry (MS) for identifying and quantifying the molecular weight of each separated component. The GC separates the mixture into individual components, allowing them to enter the mass spectrometer one at a time. This ensures that the molecular mass and fragment mass values are determined for each component individually. (Dr. Naseem Ahmed)

In another study carried out by Ahmad Aqe et al., indicates that Forensic analysis of fire debris often employs solid-phase micro-extraction followed by gas chromatography-mass spectrometry to detect components of gasoline and diesel fuel. The findings reveal that gasoline and diesel fuel accelerants exhibit greater persistence on wool and silk compared to other substrates. This underscores the importance of understanding fuel persistence times post-extinguishment and highlights wool and silk as optimal materials for scanning for ignitable liquids at fire scenes. This information is crucial for forensic experts as it helps in identifying the presence of accelerants and understanding the behavior of fuels after a fire has been extinguished. (Ahmad Aqe,2016)

A study carried out by Niculina-Sonia Suva et al., First responders use field-portable gas chromatograph-mass spectrometers (GC-MS) to analyze a variety of chemical compounds, including drugs, explosives, toxic industrial substances, chemical warfare agents, and other hazardous materials. These portable devices can measure gases, volatile and semi-volatile liquids, and vapor emissions from certain solids. Their portability allows for quick deployment in the field, enabling the identification of hazardous substances during incident responses. This facilitates real-time decision-making for risk mitigation and management, leading to more robust safety measures and effective incident resolution. Controlled field testing has shown that portable GC-MS can provide preliminary analytical data on volatile organic chemicals detected in air samples from both active and extinguished fires. (Niculina-Sonia Suva,2023)

Another research carried out by Abdulrhman M. Dhabbah et al., the collected samples were then analyzed using GC-MS to identify the presence of gasoline residues. The SPME fiber was inserted into the GC-MS injection port for 100 seconds for desorption and quantification of the analytes. The analysis revealed that after burning textiles, no detectable compounds were found after two hours. The persistence of gasoline residues was longer on synthetic fabrics (nylon and polyester) than on natural materials (cotton and wool). This method proved to be simple and efficient, making it useful for many forensic applications.(Abdulrhman M. Dhabbah, 2023)

SPME

Solid-Phase Microextraction (SPME) operates on the principle of partitioning, where the analyte reaches equilibrium between the sample matrix and the polymeric phase on the fiber. The Nernst distribution law governs the amount of analyte extracted by the coated surface. For a liquid polymeric coating like polydimethylsiloxane (PDMS), the amount of analyte absorbed at equilibrium is directly proportional to its concentration in the sample.

Another research carried out by Ahmad Aqe et al., indicates In this investigation, a polydimethylsiloxane SPME fiber was used to extract gasoline and diesel accelerants before their separation and analysis via GC-MS. The primary goal was to assess the evaporation rates and persistence of these accelerants on different substrates. The SPME fiber, composed of fused silica coated with various extraction phases, was introduced into the injection port of a gas or liquid chromatograph for desorption and analysis of the extracted analytes. This method helps in understanding how long accelerants remain detectable on different materials after a fire. (Ahmad Aqe, 2016)

In a study carried out by Abdulrhman M. Dhabbah et al., the SPME fiber, which is constructed from fused silica and coated with multiple extraction phases, was employed to retrieve the residual fire accelerant from fabric samples. Various substrate materials, including wool, cotton, silk, and polyester, were infused with 200 μ L of the accelerant and ignited for 2 minutes before the flames were extinguished. Fire debris samples were collected in nylon bags to mitigate the evaporation of volatile compounds. The SPME fiber was then placed in direct contact with the fire debris for 1 minute to facilitate the extraction of accelerants. This methodology enabled the collection of a wide array of characteristic gasoline constituents, particularly aromatic hydrocarbons, from the different non-burnt textile samples for as long as 4 hours (Abdulrhman M. Dhabbah,2024)

In another research study carried out by Arnon Grafit et al., Solid-Phase Microextraction (SPME) fiber protector is a device designed to address the fragility of the SPME fiber. It is easily assembled on the SPME device and can be removed by unscrewing for sampling to the injector. The SPME with the fiber protector was tested using headspace SPME (HS-SPME) for gasoline and diesel fuel vapor analyses. The results showed an enhancement in the detection of lighter hydrocarbons with the protector, without altering the method sensitivity. The protector can be used for years, keeping the fibers intact for hundreds of samplings. (Arnon Grafit)

SERS

Surface-enhanced Raman Scattering (SERS) is a phenomenon that **enhances the Raman scattering signals of molecules close to nanostructured metallic surfaces**, typically gold or silver. Indeed, these surfaces create intense local electromagnetic fields, amplifying the Raman signals of nearby molecules.

In a research study carried out by Cyril Muehlethaler et al., discusses surface-enhanced Raman spectroscopy (SERS), a technique that greatly boosts Raman signal strength when a specimen is placed on rough metal or semiconductor surfaces. It highlights the ability to achieve enhancement factors as high as 10^{12} due to interactions between the molecule and the substrate. This signal amplification is linked to three main resonances: electromagnetic effects involving surface plasmon resonance (SPR), molecular resonance in the molecule, and charge-transfer resonances between the molecule and the substrate. The choice of excitation wavelength can optimize this molecular resonance, leading to surface enhanced resonance Raman spectroscopy (SERRS). Silver colloids are the most common SERS substrates, known for their stability and effective SPR in the visible region, which enhances the clarity of the spectra. Adjusting particle size and shape can further improve the useful spectral range. (Cyril Muehlethaler, 2016).

Raman spectroscopy

Unlike FTIR Spectroscopy that looks at changes in dipole moments, Raman looks at changes in a molecular bonds polarizability. Interaction of light with a molecule can induce a deformation of its electron cloud. This deformation is known as a change in polarizability. Molecular bonds have specific energy transitions in which a change of polarizability occurs, giving rise to Raman active modes. As an example, molecules that contain bonds between homonuclear atoms such as carbon-carbon, sulfur-sulfur, and nitrogen-nitrogen bonds undergo a change in polarizability when photons interact with them. These are examples of bonds that give rise to Raman active spectral bands but would not be seen or difficult to see in FTIR.

Because Raman is an inherently weak effect, the optical components of a [Raman Spectrometer](#) must be well-matched and optimized. Also, since organic molecules may have a greater tendency to fluoresce when shorter wavelength radiation is used, longer wavelength monochromatic excitation sources, such as solid-state laser diodes that produce light at 785 nm, are typically used.

In a research carried out by Vijay Kumar Yadav et al., Vibrational spectroscopy, including infrared absorption and Raman scattering, is a key non-destructive method for examining evidence. While most research has focused on identifying common household substances, there is growing interest in using these techniques to identify ignitable liquids in arson investigations. The article reviews studies on fire debris and accelerants using vibrational spectroscopy, highlighting its potential and discussing limitations and future perspectives. Further research and development are needed to fully realize its potential in forensic science. (Vijay Kumar Yadav,2020)

Another research study carried out by Arunrat Cheenmatchaya et al., Transforming Raman spectra into image representations using continuous wavelet transformation (CWT) and employing transfer learning for classification is a novel approach. Achieving 73% and 53% accuracy for 50% and 25% weathered gasoline samples, respectively, is impressive. This method effectively processes and classifies complex Raman spectral data without the need for feature engineering, making it a promising tool for enhancing crime scene intelligence based on chemical signatures identified by portable Raman spectrometers. (Arunrat Cheenmatchaya, 2016)

Metallic Residues

X-Ray Fluorescence (XRF)

X-Ray Fluorescence has become the one of the most versatile analytical method for determining the elements from electronic waste like printed circuit boards ,lithium ion batteries and LEDs .For solid sample analysis it is the most cost efficient and rapid tool compared with inductively coupled plasma mass spectrometry and instrumental neutral activation analysis.

XRF is a non-destructive analysis which does not require any complex sample digestion and provides faster results but the lack of matrix-specific calibration standards limits the quantification accuracy of certain elements and even it is effective for major components it is not suitable for light elements like aluminum, silicon. Reliable results require particle sizes below 200 μm and larger particles increase variability. Portable XRF has higher variability and detection limits compared to wavelength dispersive XRF.

XRF is a promising analytical method for recycling industries, but calibration and detection limitations need to be addressed for full potential in resource recovery and sustainable waste management.(ShaunT.Lancaster,2024)

Femtosecond laser ablation-ICP-mass spectrometry

Vanhaecke et al.'s paper presents a sophisticated analytical approach using femtosecond laser ablation-ICP-mass spectrometry (fs-LA-ICP-MS) to determine platinum group metals (PGMs) and gold concentrations in lead fire-assay buttons. PGMs and gold are critical materials used in catalysts, electronics, and jewelry due to their high value and limited availability. Challenges include low PGM concentrations in ores, complex matrices, and limitations of earlier methods like nanosecond LA-ICP-MS.

Femtosecond LA-ICP-MS offers advantages such as controlled ablation, improved accuracy, and better detection limits. Its experimental design includes a laser with a wavelength of 795 nm, pulse duration of 150 fs, energy of 2.4 mJ, repetition rate of 20 Hz, beam diameter of 150 μm , carrier gas of argon, dynamic reaction cell of ammonia gas, and internal standard of the isotope $^{204}\text{Pb}^+$

The results show that all experimental results for Rh, Pd, Ru, and Pt matched certified values within 20%, with most within 10%. Rh and Ru determination benefited from interference-free conditions provided by the reaction cell. Detection limits ranged from 0.003 mg/g for Rh to 0.015 mg/g for Ru, significantly lower than those from nanosecond LA-ICP-MS.

Femtosecond systems demonstrated superior detection power, better quantitative agreement with certified values, reduced biases, and standard deviations in measured values. The method proved suitable for high-throughput applications, with sample analysis times of 10-15 minutes per button. Matrix-matched calibration standards ensured high accuracy and precision, even for trace elements like Ir and Au. (Frank Vanhaecke, 2010)

The study aims to analyze the oxidation behavior of carbon steel in a simulated fire environment, specifically in an ethanol-combustion atmosphere. The research uses a laboratory device to simulate an ethanol-combustion atmosphere and exposes carbon steel samples to varying oxidation durations. Surface morphology, phase composition, and visual analysis techniques are used to analyze the results. Key observations include mesh-like oxide patterns on carbon steel, which form due to surface defects and expand over time. The study also reveals that incomplete combustion of ethanol leads to the occurrence of elemental carbon. Oxidation products include wustite, magnetite, and hematite, which are influenced by temperature and oxidation duration. The findings provide forensic evidence to identify accelerants in arson investigations. (Hao Hong, 2020)

AAS

The paper discusses a method for determining firing distances in forensic investigations using Atomic Absorption Spectroscopy (AAS). The study involved shooting a Colt 38 Special at various distances (5 to 100 cm) and examining lead concentrations in three concentric rings cut from the target. Key findings include the target design, lead analysis, and methodology. The first ring captures the entrance hole and bullet wiping rim, while the second and third rings capture smoke halos and firearm discharge residues. The results showed a semi-logarithmic relationship between firing distance and lead concentration, with lead amounts decreasing with distance. The study emphasizes the importance of accurate sampling and calibration in determining firing distances and outlines specific procedures for collecting samples and performing quantitative analyses. The findings can help distinguish between homicide and suicide in shooting cases, particularly at close ranges.

Gamma-ray fluorescence spectrometer

A study conducted by Tomasz Mach and colleagues analyzed the elemental composition of PM10-bound metals in a fire station garage in Poland. The researchers used a gamma-ray fluorescence spectrometer to measure the elemental composition of particulate matter, collecting data every four hours. They employed three models for source apportionment: Principal Component Analysis (PCA), EPA UNMIX, and EPA Positive Matrix Factorization (PMF).

Key findings include significant temporal variations in the concentrations of PM10 and the studied metals, indicating that pollutant levels can change rapidly based on activities within the firehouse. Model comparisons showed that both UNMIX and PMF models provided more interpretable results by constraining coefficients to non-negative values. The study identified four main components contributing to the PM10 levels, with the positive constraints of the UNMIX model leading to more physically interpretable source profiles.

The research highlights the importance of high-resolution measurements in understanding indoor air quality in specific environments like fire stations, where pollutant sources can vary significantly over short periods. The findings contribute to better environmental management and protection strategies in such unique settings.

The study was conducted over a two-week period from June 24 to July 8, 2020, in a residential area of Warsaw. The researchers applied three different receptor models: Principal Component Analysis (PCA), EPA UNMIX, and EPA Positive Matrix Factorization (PMF).

Findings revealed high levels of certain metals, particularly sulfur, zinc, arsenic, nickel, cadmium, and lead, indicating significant anthropogenic influences. Combustion processes, particularly from fire engines, significantly contributed to the PM10 levels in the garage, highlighting the role of operational activities in air quality degradation. Other identified sources included mineral dust and

road dust, which were linked to both resuspension and abrasion processes occurring inside the firehouse and from external environments.

A study by Diauddin R. Nammari and colleagues examined the environmental and safety implications of controlled fires in municipal solid waste (MSW) bales for energy production. The research focused on the flammability, combustion processes, and emissions resulting from burning the bales, which are tightly wrapped in polyethylene to minimize self-ignition risks. The experiments simulate the effects of large-scale fires in full-scale bale storage areas. Despite the high moisture content and density of the bales, the results indicate that the bales combust effectively when ignited without significant risks of spontaneous ignition. Key parameters measured during combustion included concentrations of various gases, smoke gas rates, and temperatures, as well as the collection and analysis of soot particles for heavy metals and organic pollutants.

The study found varying levels of dioxins and polycyclic aromatic hydrocarbons during combustion, with concentrations sometimes exceeding European Union (EU) emission limits. Notably, HCl levels were particularly high, exceeding limits by tenfold. The findings suggest that while baling is an effective method for waste fuel storage, the potential for high emissions during combustion requires further investigation and monitoring to understand the environmental and health risks associated with full-scale storage fires, especially in proximity to urban areas.

Inductively coupled plasma optical emission spectrometry (ICP-OES)

A study by Jen-Hao Chi and Peng-Chi Peng published in the Journal of the Chinese Institute of Engineers aimed to determine the cause of an arson fire at a factory in Taoyuan County, Taiwan. The investigation used advanced techniques such as metallographic analysis, inductively coupled plasma optical emission spectrometry (ICP-OES), and thermal analysis to analyze debris from the fire scene. The fire occurred on June 12, 2012, at a factory with a significant amount of pharmaceutical products stored. The fire resulted in substantial damage, and the factory owner had recently increased fire insurance due to flooding concerns. Initial investigations revealed no ignitable liquid residues, prompting further analysis to rule out potential causes. The researchers identified polyurethane (PU) foam, a highly flammable material used in the factory's refrigerators, as a critical factor in the fire's rapid spread. Their findings demonstrated that PU foam begins thermal decomposition at approximately 130°C, releasing significant heat that can exacerbate fire conditions.

Through detailed examination of various samples and multiple testing techniques, the researchers concluded that the fire was likely a result of arson rather than accidental ignition. The study highlights the importance of comprehensive fire investigation techniques and the need for a multidisciplinary approach to accurately determine fire causes. The results are not only significant for understanding this specific incident but also serve as a reference for future arson investigations, emphasizing the challenges faced by forensic investigators in the field.

Inductively coupled plasma mass spectrometry (ICP-MS)

The study by R. Juvonen, T. Lakomaa, and L. Soikkeli from the Geological Survey of Finland focuses on the challenges of determining gold and platinum group elements in geological samples, particularly using the nickel sulphide fire assay (NiS-FA) method followed by inductively coupled plasma mass spectrometry (ICP-MS). The authors highlight the low average concentration of Au and PGEs in rocks, often below 3 ng/g, which necessitates larger sample sizes for effective pre-concentration before instrumental analysis. The NiS-FA technique has gained popularity since its introduction in the 1970s and is currently the preferred method for analyzing these elements in geological samples. However, specific sample matrices pose challenges during the fusion process, leading to incomplete separation of the NiS button from the slag or the formation of mixed sulphide buttons. The study identifies several strategies to overcome these challenges, such as mitigating interference caused by graphite in black shale samples, adding a reducing agent to magnetite samples, and correcting interference of Cu on rhodium determination in ICP-MS. The research acknowledges the lack of reference materials for difficult-to-fuse samples and suggests varying fusion reagents based on sample matrices to enhance outcomes.

In the 1950s and 1960s, extracts were analyzed using infrared (IR) or ultraviolet (UV) spectroscopy, but these techniques were neither very sensitive nor specific. Gas chromatography (GC) became the analytical method of choice in the late 1960s, using pattern recognition techniques to isolate smaller quantities of ignitable liquid residue. GC columns were glass or metal tubes, 1/4 in. in diameter by 6 to 10 ft long. However, there were problems with these early capillary columns, and as late as 1990, packed columns were contemplated in ASTM E1387, Standard Test Method for Flammable or Combustible Liquid Residues in Extracts from Fire Debris Samples by Gas Chromatography.

Flame ionization detection (FID) improved the sensitivity of GC detectors by a couple of orders of magnitude over TC detectors. Some laboratories were using gas chromatography–mass spectrometry (GC–MS) as early as 1976, but mass spectrometers were expensive, not reliable, and required a computer. By the early 1980s, mass spectrometry was still expensive but became more widespread in fire debris analysis, particularly in betterfunded laboratories. Public laboratories acquired GC–MS instruments for use in drug identification, and this was another reason that they became available for fire debris analysis.

The instrumentation of the 1980s was more automated and could collect a mass spectrum several times per second, even if no peak was eluting. This resulted in a much more efficient process, but data files were very large. Today's GC–MS collects data every tenth of a second, using sophisticated software to keep the file size to around a megabyte.

In 1982, Martin Smith published an article about a technique called mass chromatography, which utilized a computer to separate the mass spectral signals according to the functional groups of the compounds that produced them. This powerful analytical tool is known as extracted ion profiling or extracted ion chromatography and forms the basis of most identifications. The development of the personal computer made it possible for average laboratories to control a mass spectrometer, and instrument manufacturers responded by producing benchtop models with increasing sensitivity and extraordinary robustness.

Trace Residues

What is trace evidence with respect to arson cases? Trace evidence can include anything from fibers, hairs, glass pieces, gunshot residues, surface coatings, cosmetics, soil, wood chips, to residues from ignitable liquids. Trace evidence provides crucial evidence in the field of forensic science. Once identified, samples must be collected using proper protocols to prevent contamination and ensure proper evidence integrity. Once collected, samples can be analyzed using various scientific techniques, including microscopy, chromatography, spectroscopy, etc. They help in analyzing the origin and composition of the evidence and also help in providing a link between the suspect and the crime scene.

Another important technique that helps in arson investigation is analyzing fire debris which mainly involves analyzing charred debris and identifying the presence of accelerants used. Various other trace evidences such as fingerprints, biological remains, DNA, toolmarks, etc. can also be analyzed through a variety of cutting-edge technologies.

Some of the commonly used analytical techniques for trace evidence analysis include Gas Chromatography-Mass Spectrometry (GC-MS), Laser Ablation Inductively Coupled Plasma Mass Spectrometry (LA-ICP-MS), Surface-Enhanced Raman Spectroscopy (SERS), Scanning Electron Microscopy-Energy Dispersive X-ray Spectroscopy (SEM-EDS), Transmission Electron Microscopy (TEM), Ion Chromatography, Fourier-transform infrared (FTIR) Spectroscopy and X-ray diffraction (XRD). Some of which have already been mentioned in the previous residues.

Microscopic techniques can also be used to analyze certain residues such as hair, fibre, paint, or any other debris found at the scene of crime. Microscopic techniques can be coupled with spectroscopic techniques for elemental analysis and origin identification.

Chromatography

Gas chromatography (GC) combined with mass spectrometry (MS) is the preferred method for the analysis of ignitable liquid residues (ILRs) in fire debris. The American Society for Testing and Materials (ASTM) provides guidelines for the identification of ILRs by GC-MS (ASTM standard E1618). In a paper by Marie Martin Fabritius et al. describes a method for the analysis of volatile compounds in fire debris using ACS and automated thermal desorption-gas chromatography-mass spectrometry (ATD/GC-MS). In this process, ACS is embedded in the debris where volatile substances are trapped; this is followed by a thermal desorption that has the function of releasing the trapped substances. The volatiles released are then analyzed by GC-MS, separating and identifying the chemicals.

A wide variety of flammable liquids, consisting of highly volatile (methylated alcohol, universal diluter), medium volatile (petroleum distillate, brush cleaner, lamp oil), and low volatile compounds (candle wax), were determined by both ACS ATD-GC-MS and ACS GC-MS in this study by Marie Martin et al. Detection is possible by desorption of ACS even for extremely light compounds, such as ethanol, which, however, was not possible due to the delay in solvent effect in the pentane desorption process for its analysis by GC-MS. Extremely light compounds evaporate quite rapidly under conditions prevailing in real fires and may thus not have been identified within any of the fire debris samples. However, it is important to detect these light compounds for comparison with unburned samples used as reference material, such as unburned wood, soil or clothes of suspected arsonists. (Martin Fabritius et al., 2018)

GC is available in various modes, which include GC-flame ionisation detector (GC-FID), GC-photo ionisation detection (GC-PID), GC-mass spectroscopy (GC-MS), GC-differential mobility spectrometry, GC-tandem mass spectroscopy and multidimensional GC. The technique has many advantages, but at the same time, it also has some limitations. It is a destructive method of analysis. Presently, the scientific fraternity is inclined towards non-destructive methods. (Yadav et al., 2020)

Spectroscopy

Due to their sensitivity, non-destructive spectroscopic techniques are emerging as the best analytical tools for evidence analysis since, in most cases, the sample may be found in trace amounts.

IR and Raman spectroscopy has become commonly used in forensic science over the past few years because they can withstand analytical requirements for the analysis of evidence. The two spectroscopic techniques combined are exclusively known as vibrational spectroscopy. The molecular vibrations provide characteristic information on the structure of a substance. IR and Raman spectroscopy can reveal the structure of molecules to identify any substance. The level of evidence damage is so high in arson cases that confirmatory analysis of accelerant remains and fire-damaged materials is quite challenging to carry out with a degree of sufficient accuracy. Figures 1 and 2 depict the schematic representation of IR and Raman spectroscopy instrumentation. Vibrational spectroscopic techniques are the most appropriate for the analysis of arson-related samples because in most cases there is a trace amount of the sample, while the possibility of further damage is also high during the analytical process.

More studies are based on Raman spectroscopy rather than NIR and ATR-FTIR. More studies are required to strengthen the technology for further application in arson investigation.

Laser Induced Breakdown Spectroscopy

Another research carried out by Soojin Choi et al., indicates that LIBS has capabilities for in-situ analysis and depth profiling hence a possibility of laser-induced breakdown spectroscopy in fire investigation. LIBS is an atomic emission spectroscopy based on plasma. Plasma is formed when the high-power pulsed laser focused onto the surface of a sample breaks its chemical bonds. Information on electrons, ions, atoms, and molecules can be retrieved from the emitted plasma signal. The LIBS analysis does not require sample extraction or preparation and allows for real-time in-situ analysis at a fire scene. Depth profiling with LIBS is able to determine the degree of carbonization of combusted samples. LIBS analysis cannot substitute for conventional analytical methods such as gas chromatography but can be helpful in assisting fire investigation at crime scenes by data analysis and in-situ forensic signal detection.(Choi & Yoh, 2017)

Fingerprint Recovery

There are 5 major stages to a fire; Pre-heat, Early growth, Flashover, Steady state and Decay. It is very important to know these stages because not only will each stage have a different adverse effect on fingerprints and DNA, but also cause physical changes to the surfaces that they adhere to.

By sticking to the surface of fingerprints and DNA matrices, soot can stop them from evaporating. Even with the help of various light emissions, latent finger traces and blood staining are not always evident since soot deposition is frequently dense and dark in appearance. Therefore, in an effort to protect the possible evidence underneath, methods for removing soot in the least harmful way have been created. Light brushing is mostly utilized as a quick and simple way to remove unwanted soot from both porous and non-porous surfaces. Lifting tape would next be used to smooth, non-porous surfaces, followed by either sodium hydroxide (which is harmful to DNA) if the soot deposition is heavy or an eraser for burnt fingerprints. The Mikrosil method would be applied if the surface has texture. Liquid latex would also be utilized if a sizable area needed to be coated.

Soot Removal Techniques:

Sl. No.	Method used :	Procedure :	Evidence Type :
1.	2% Sodium Hydroxide	NaOH is mixed with water and applied through a spray bottle.	Useful for Fingerprints.
2.	2% Sulfo-salicyclic acid	Sprayed using a spray bottle to fix the latent ridges onto the non porous surfaces before being placed onto a sonic bath	-Fingerprints
3.	silicone rubber casting (Mikrosil)	Mikrosil is mixed with the Mikrosil hardener, this paste is then applied to the surface and allowed to set before being removed.	- Fingerprints
4.	Ultrasonic bath	Contains water, gasoline, toluene, xylene, chloroform, ethanol, acetone, hexane, diluted sulfuric acid and detergent	Fingerprints & DNA
5.	Liquid latex	Liquid latex along with a thickening agent and colourant can be sprayed through a spray gun with a suppressor onto surfaces, only dried can be peeled off, removing the soot	Fingerprints & DNA
6.	Manual techniques such as using an Eraser, Tape lifting and Light Brushing	1.light rubbing of the surface with the eraser on metal surfaces. 2.Lifting fingerprints with Tape can help collect residues 3. Allows carbon to adhere to the oil residue within the print using a fiberglass fingerprint brush	Fingerprint

Different means of enhancement techniques are required for Patent, Latent and Plastic prints. Visualization Techniques such as Ninhydrin, Superglue fuming method, Physical developers and DFO, Small Particle Reagent (SPR), etc. are used.

In 2013, a novel method for the development of Latent Fingerprints was developed, which talks about Zinc carbonate based fluorescent small particle reagent (SPR) was capable of developing latent fingerprints exposed to a maximum temperature of 800°C. (Dhall et al., 2013)

Formulations A and B were created in SPR to test out the differences caused in latent fingerprints. Formulation A is comprised of a suspension of basic zinc carbonate in distilled water, to which eosin B and commercial liquid detergent were added. To formulation B, in a suspension of zinc carbonate in distilled water, eosin Y and commercial liquid detergent were added. Fingerprints were then illuminated with radiation having 505–550 nm wavelength. When observed through orange goggles, the fingerprint exhibited green fluorescence. (Dhall et al., 2013)

DNA Recovery

Leicestershire Police were looking into the 1986 rape and killing of Dawn Ashworth and Linda Mann, two schoolgirls. This was the first forensic inquiry to use genetic profiling and DNA fingerprinting. Blood samples seized from suspect Richard Buckland, a learning-disabled teen who confessed to the murder, were compared to semen samples recovered from the girls. Richard Buckland was exonerated of their murder because the DNA profiles from the two semen samples matched one another but not his. 4,000 nearby guys who had no alibi for the crime had samples of their blood and saliva found. Colin Pitchfork was eventually found guilty of two murder charges. (Channel4 News. Five things we know from 30 years of DNA fingerprinting. 2014)

The foundation of all life is DNA. The genetic substance that gives each individual their unique identity is called DNA. The DNA evidence that will produce a complete DNA profile is the most valuable. Degraded DNA is less likely to provide a complete profile. Since both the fire and its extinguishment might damage DNA, it's critical to comprehend how arson affects various DNA samples. When samples are exposed to arson, DNA deterioration can take many different ways. Phosphodiester bonds may undergo hydrolytic breakage as a result of the extremely high temperatures generated.

More short loci have been discovered in degraded DNA profiles, which is especially helpful when examining Short Tandem Repeat Polymorphism (STR). A technique for screening samples prior to typing has been developed. Specifically, for evidence that is known to have been subjected to harmful conditions, resulting in low DNA quality, and that is of crucial value. Based on the frequent and plentiful Alu sequences, this DNA degradation detection assay can determine the amount of viable DNA. DNA degradation can be quantitatively measured using the ratio of the two fluorophores. Severe DNA degradation is indicated by a ratio of 2 or greater; the higher the ratio, the more degraded the DNA. The Taqman probe was utilized in the fluorescence-based technique to attach to a particular DNA sequence.

Different chemical compounds (dyes) that emit light at different wavelengths will be present in the probes that are affixed to the blood and semen. Fluorescence at that wavelength increases with the amount of DNA present. The goal was to determine whether the amplicon's semen to blood ratio varied in any way. The deterioration of semen to blood samples following exposure to the intense heat of a fire did not differ significantly, according to the data. This experiment did, however, demonstrate how various substrates can impact the caliber of biological evidence. (O'Hagan & Calder, 2020)

In a study conducted by Vineyard A et al., blood samples exposed to various arson scene-related factors were identified using several blood detection reagents. Various blood dilutions were applied to wood blocks, and extinguishment techniques such as smothering and water covering were used. Half of the samples were mixed with unleaded gasoline to test the effects of accelerant on detection techniques. Bluestar, Luminol, and phenolphthalein tetramethylbenzidine (PTMB) were among the blood testing techniques used. The capacity to identify blood using all three detection modalities is adversely affected when fire is put out with water. When utilizing Luminol or Bluestar, positive blood results from porous combustible surfaces are more likely to be successful. Vineyard A et al. came to the conclusion that presumptive tests are insufficiently sensitive to identify DNA samples on burned objects, despite the porous surface only being exposed to the fire for a minute. (Vineyard et al., 2019)

Depending on the scene and the fire's factors, different enhancing techniques and blood-detecting procedures perform well on various substrates and have varying success rates.

Several DNA enhancement techniques were introduced. Some of the common methods include Bluestar, Kastle Meyer, Benzidine, PCR for amplification of DNA, Amido black, Leucocrystal Violet, Phadebas test, ABA Hematrace Card, RSID, Acid Phosphatase test, Christmas tree stain, and various other Alternate light sources. (O'Hagan & Calder, 2020)

Latent Bloodstains

A modified digital single-lens reflex (SLR) camera was used in a different 2018 investigation to examine the use of reflected infrared photography to find latent bloodstain evidence under different deposited overlying soot densities. A number of steps made up this experimental design, including (i) modeling the fire, (ii) taking pictures and imaging them, (iii) measuring the soot density at different hot layers and wavelengths, (iv) looking at the pictures, and (v) analyzing the findings. Utilizing calibrated exposure with constant lighting and relative exposure values, experimental samples were captured on camera utilizing reflected infrared and conventional visible light photography techniques. To ascertain whether infrared was a useful

forensic imaging technique for detecting bloodstains behind thick soot deposits from a fire, a number of factors were investigated. (Bastide et al., 2019)

Under thick layers of soot, latent bloodstains could be successfully found using reflected infrared imaging. This was accomplished by combining the spectral responses of four parameters that are required for this technique: (i) the increased infrared absorption properties of blood following heat exposure; (ii) the soot layer's infrared transmission and penetration; (iii) the painted background's infrared reflection; and (iv) the digital camera's sensitivity and the light source's spectral output. Latent bloodstains may become evident as a result of the combination of these distinct infrared responses of blood, soot, and background, which results in a difference in the tonal values or contrast recorded between bloodstains and overlaying soot.

The qualitative comparison of photos taken using reflected infrared photography and ordinary visible light photography showed that the infrared methods could identify bloodstains underneath thick layers of soot (and penetrate the soot layer). Additional quantitative and statistical analysis supported this qualitative finding. Pixel brightness profiles of the photos were created using image analysis, and the ability to visually depict the clarity of bloodstains after applying increasing soot densities was demonstrated graphically. (Bastide et al., 2019)

Conclusion

Arson investigation is an intricate and multifaceted discipline requiring integrating crime scene techniques, analytical methods, and evidence management to ensure accurate and reliable outcomes. Advances in technology, such as digital forensics, fire pattern analysis, and innovative analytical instruments like GC-MS and SPME, have significantly improved investigators' ability to detect accelerants and interpret fire dynamics. This, coupled with rigorous evidence collection protocols and chain of custody practices, ensures the integrity and admissibility of findings in legal contexts.

Understanding fire behavior and dynamics, including mechanisms of heat transfer and combustion phases, is fundamental to distinguish between accidental and intentional fires. Emerging methods like ATIR-FTIR and Raman spectroscopy enable non-destructive analysis of fire residues. At the same time, LIBS and SERS offer real-time, sensitive detection capabilities, further enhancing the precision of investigations.

The integration of forensic entomology, advanced fingerprint recovery techniques, and DNA analysis underscores the importance of multidisciplinary approaches in addressing arson cases. These methods not only improve the identification of suspects but also provide critical insights into the sequence and cause of events.

While significant progress has been made, challenges remain, including the need for more robust non-destructive techniques, better persistence analysis of accelerants, and enhanced methodologies for detecting latent evidence under extreme conditions. Future research and development should focus on optimizing these techniques, fostering interdisciplinary collaboration, and ensuring that investigators are equipped with cutting-edge tools to combat this complex crime effectively.

By building on these advancements, arson investigations can continue to evolve as a critical component of forensic science, aiding law enforcement in achieving justice and mitigating the profound societal impact of this destructive crime

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