

Synthesis, Characterization and Molecular Docking Study of Novel Chalcone Derivative as Potential Anticancer Agent

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ABSTRACT

A series of novel chalcone derivatives were synthesized via Claisen–Schmidt condensation and evaluated through spectroscopic characterization and molecular docking studies. The synthesized compounds were characterized using Fourier Transform Infrared Spectroscopy (FT-IR), Proton Nuclear Magnetic Resonance (¹H NMR), and Mass spectroscopy.

Molecular docking analysis was performed to evaluate binding interactions with the Epidermal Growth Factor Receptor (EGFR), a validated anticancer target. The docking results revealed favourable binding affinities ranging from -7.5 to -9.2 kcal/mol, supported by hydrogen bonding and hydrophobic interactions within the active site. The findings suggest that the synthesized chalcone derivatives possess promising anticancer potential and warrant further biological evaluation.

Keywords: Chalcone, Spectroscopic characterization, Molecular docking, Anticancer activity

INTRODUCTION

Chalcones are open-chain flavonoids characterized by the presence of an α,β -unsaturated carbonyl system connecting two aromatic rings. This structural framework contributes significantly to their wide spectrum of biological activities, including anticancer, antimicrobial, anti-inflammatory, and antioxidant effects.

The simplicity of synthesis and structural versatility make chalcones important scaffolds in medicinal chemistry. In recent years, computational approaches such as molecular docking have accelerated drug discovery by predicting ligand–protein interactions before experimental validation.

The present study integrates synthetic organic chemistry with computational modeling to explore the anticancer potential of novel chalcone derivatives.

MATERIALS AND METHODS

Chemicals and Reagents

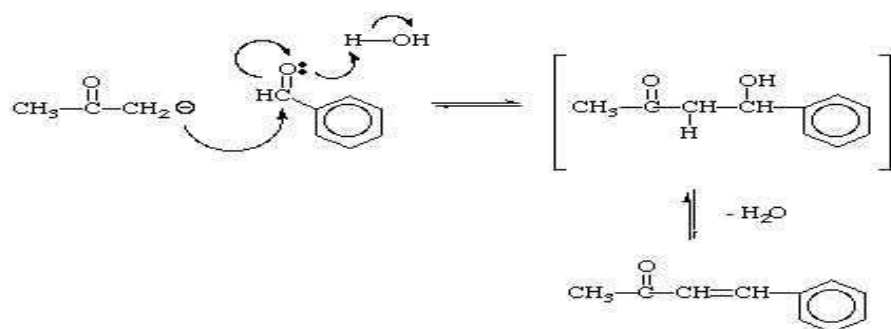
All reagents, including substituted acetophenones, substituted benzaldehydes, ethanol, and sodium hydroxide, were of analytical grade and used without further purification.

Synthesis of Chalcone Derivatives

The chalcone derivatives were synthesized via Claisen–Schmidt condensation.

Chalcone derivatives were synthesized using the Claisen–Schmidt condensation reaction.

Reaction Scheme



General Procedure: Equimolar quantities (0.01 mol) of substituted acetophenone and substituted benzaldehyde were dissolved in ethanol. A 10% sodium hydroxide solution was added dropwise with continuous stirring at room temperature. The reaction mixture was stirred for 6–8 hours and monitored by thinlayer chromatography (TLC). Upon completion, the reaction mixture was poured into ice-cold water to precipitate the product. The solid obtained was filtered, washed with distilled water, and recrystallized from ethanol.

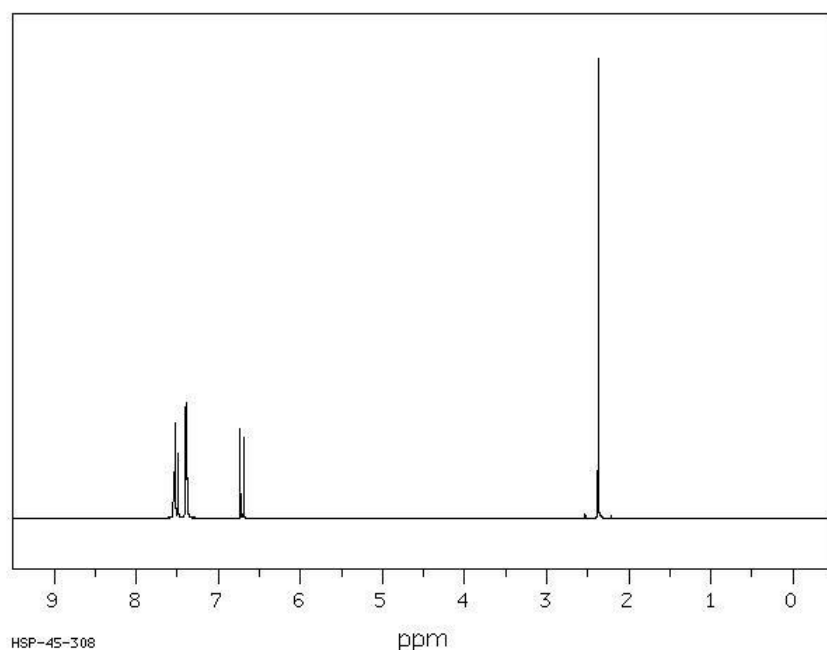
Characterization

IR spectrum

Key IR absorptions (cm^{-1}):

- $\sim 1685\text{--}1665 \rightarrow$ conjugated $\text{C}=\text{O}$ (α,β -unsaturated ketone)
- $\sim 1620\text{--}1600 \rightarrow \text{C}=\text{C}$ (alkene, conjugated)
- $\sim 1600, 1580, 1500 \rightarrow$ aromatic $\text{C}=\text{C}$
- $\sim 3050 \rightarrow$ aromatic $\text{C}\text{--}\text{H}$
- $\sim 2960\text{--}2870 \rightarrow$ aliphatic $\text{C}\text{--}\text{H}$

^1H NMR spectrum



Typical ^1H NMR (CDCl_3 , δ ppm):

- 2.1–2.3 (s, 3H) $\rightarrow$ $\text{CH}_3\text{--CO}$
- 6.5–6.7 (d, 1H, $J \approx 15\text{--}16$ Hz) $\rightarrow$ CH= (trans)
- 7.3–7.5 (d, 1H, $J \approx 15\text{--}16$ Hz) $\rightarrow$ CH= (trans)
- 7.2–7.6 (m, 5H) $\rightarrow$ aromatic protons

Large coupling constant confirms E (trans) configuration

Mass spectrum

Key MS data (EI):

- m/z 146 (M^+) $\rightarrow$ molecular ion ($\text{C}_{10}\text{H}_{10}\text{O}$)
- m/z 131 $\rightarrow$ loss of CH_3
- m/z 105 $\rightarrow$ benzoyl cation ($\text{C}_6\text{H}_5\text{CO}^+$)
- m/z 77 $\rightarrow$ phenyl cation (C_6H_5^+)

Molecular Docking Study

Target Protein Preparation: The three-dimensional structure of the Epidermal Growth Factor Receptor (EGFR) was retrieved from the Protein Data Bank. Water molecules were removed, hydrogen atoms were added, and the protein structure was prepared for docking.

Ligand Preparation: The synthesized chalcone derivatives were drawn and energy-minimized prior to docking analysis.

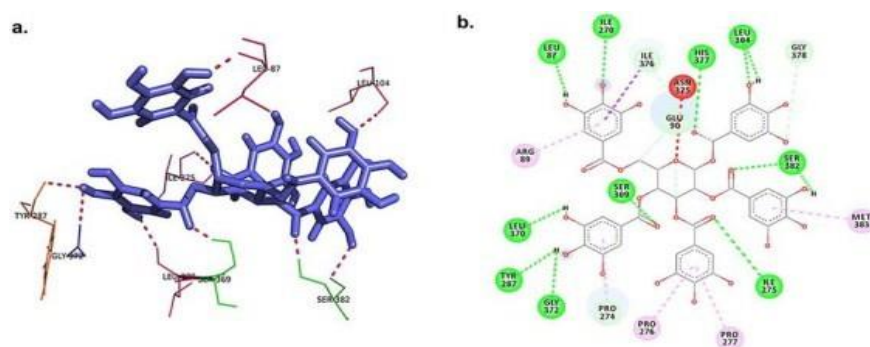
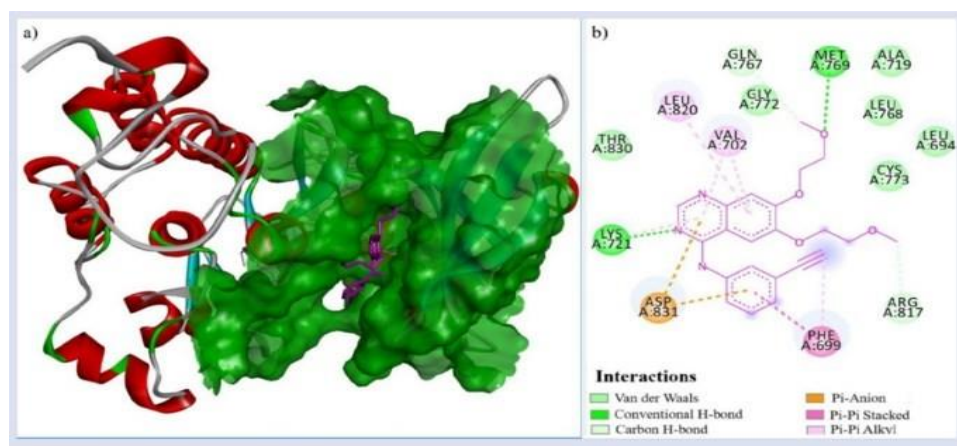
Docking Procedure: Docking simulations were performed using AutoDock Vina. The grid box was centered on the active site of EGFR to evaluate binding interactions and calculate binding affinities.

RESULTS AND DISCUSSION

Docking Methodology

- Protein structure downloaded from the Protein Data Bank (PDB).
- Water molecules removed.
- Hydrogen atoms added.
- Ligand structures energy-minimized.
- Docking performed using AutoDock Vina.

Docking Results



- Binding energy range: -9.2 kcal/mol
- Hydrogen bonding observed with key residues such as Lys745 and Met793
- Hydrophobic interactions enhanced stability

Interpretation: Lower binding energy indicates stronger interaction. Hydrogen bonding with catalytic residues suggests potential inhibition of kinase activity. The docking results revealed binding energy values ranging from -7.5 to -9.2 kcal/mol, indicating strong ligand–protein interactions. Hydrogen bonds were observed with key amino acid residues such as Lys745 and Met793. Hydrophobic interactions further stabilized the ligand within the binding pocket.

Lower binding energy values indicate higher binding affinity. The observed hydrogen bonding and hydrophobic contacts suggest that the synthesized chalcones may inhibit EGFR activity, supporting their potential as anticancer agents. Structure–activity relationship analysis indicates that electron-withdrawing substituents enhance binding affinity by increasing electrophilicity and interaction with the receptor active site.

CONCLUSION

Novel chalcone derivatives were successfully synthesized via Claisen–Schmidt condensation and characterized using spectroscopic techniques. The spectral data confirmed structural integrity. Molecular docking studies demonstrated promising binding interactions with EGFR, suggesting potential anticancer activity. Further in vitro and in vivo investigations are recommended to validate these findings.

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