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Effect of Heat Treatment on Microstructure and Hardness of Low Carbon Steel

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ABSTRACT

Low carbon steel is widely used in structural and engineering applications due to its excellent formability and weldability; however, its mechanical properties are highly dependent on microstructure. In the present study, the influence of different heat treatment processes on the microstructure and hardness of low carbon steel was investigated. Different heat treatment conditions, namely annealing, normalizing, conventional hardening (water quenching), Neem Bath ,Ice Bath,Salt Bath,brine Bath quenching were applied. Metallographic analysis was carried out using optical microscopy at magnifications of 100× and 500×. The results reveal significant variations in phase distribution, grain size, and morphology due to different cooling rates. Fine pearlitic structures were observed in normalized samples, whereas martensitic structures were dominant in quenched specimens. Hardness values increased with increasing cooling rate, with brine-quenched samples showing maximum hardness. The study establishes a clear correlation between heat treatment, microstructure, and hardness of low carbon steel.

Keywords: Heat treatment, Normalizing, Annealing

INTRODUCTION

Mechanical properties of steels are strongly dependent on their microstructure, which can be controlled through heat treatment processes. Heat treatment involves controlled heating and cooling operations to modify the internal structure of materials without altering their chemical composition.

Low carbon steel (≤0.25% C) is extensively used in construction, automotive, and manufacturing industries due to its low cost and high ductility. However, its relatively low hardness and wear resistance limit its applications in severe environments.

The transformation of austenite into ferrite, pearlite or martensite depends on cooling rate and heat treatment conditions. By selecting appropriate heat treatment techniques such as annealing, normalizing, quenching. it is possible to tailor the microstructure and thereby enhance mechanical properties.

The present work focuses on:

Studying microstructural evolution under different heat treatments Comparing hardness variations Establishing structure-property relationships

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EXPERIMENTAL METHODOLOGY

Material

Material: Low carbon steel (TMT bar)

Composition: Use Spectroscopy

Sample Preparation for Metallography

Specimens cut 1 inch in size & has diameter of 20mm.

Grinding using SiC papers (220–1000 grit)

Polishing using diamond paste

Etching using 2% Nital

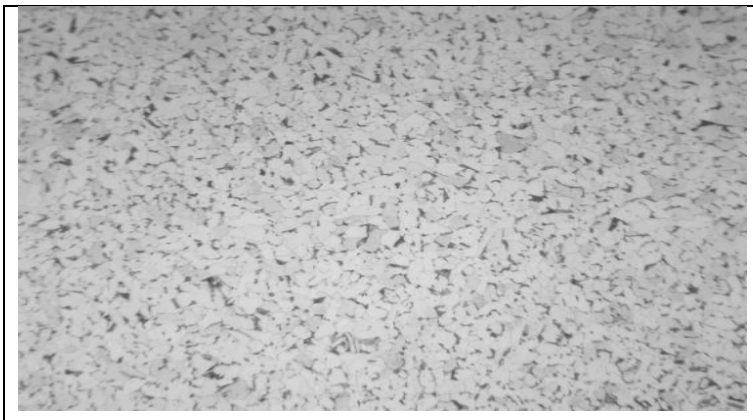
Heat Treatment Conditions

Table 1. Heat Treatment Conditions with respect to Cooling Medium

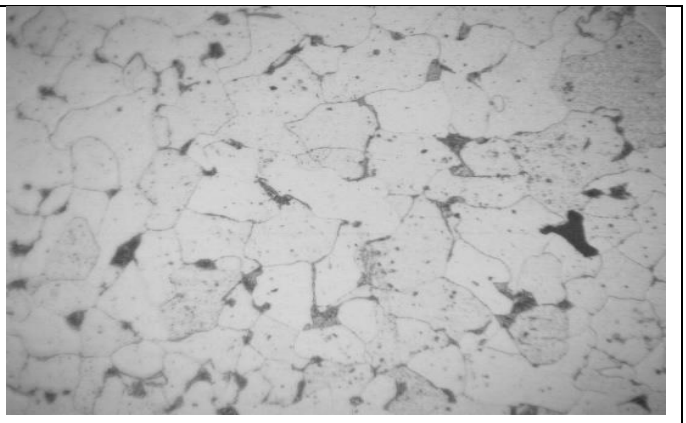
Sample No	Heat Treatment	Temperature (°C)	Holding Time (min)	Cooling Medium
1	Annealing	940	60	Furnace cool
2	Normalizing	940	60	Air cool
3	Hardening	940	60	Neem Bath (121 gm of neem in 1L of water)
4	Hardening	940	60	Ice Bath (300gm of ice in 250ml of water)
5	Hardening	940	60	Salt Bath (38gm of salt in 1L of water)
6	Hardening	940	60	Brine Solution (125gm of salt, 45gm of dish-wash, & 12.5gm of laundry detergent in 1L of water)

Microstructure Observation

Figure 1: Annealing Sample



100X



500X

Figure 2: Normalizing Sample

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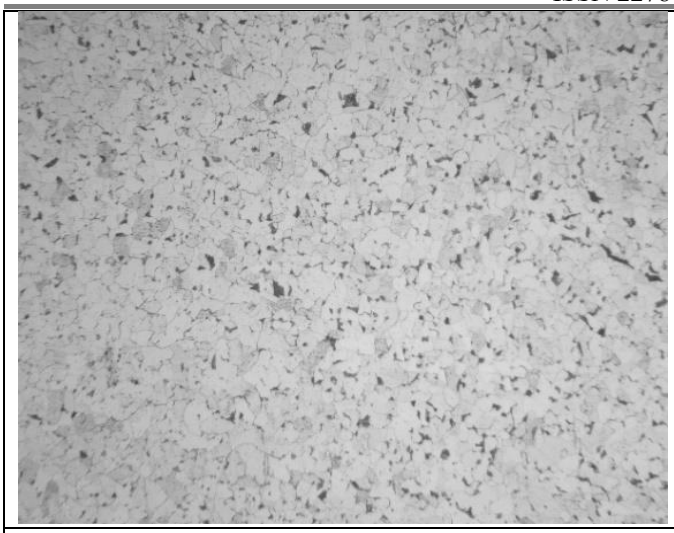
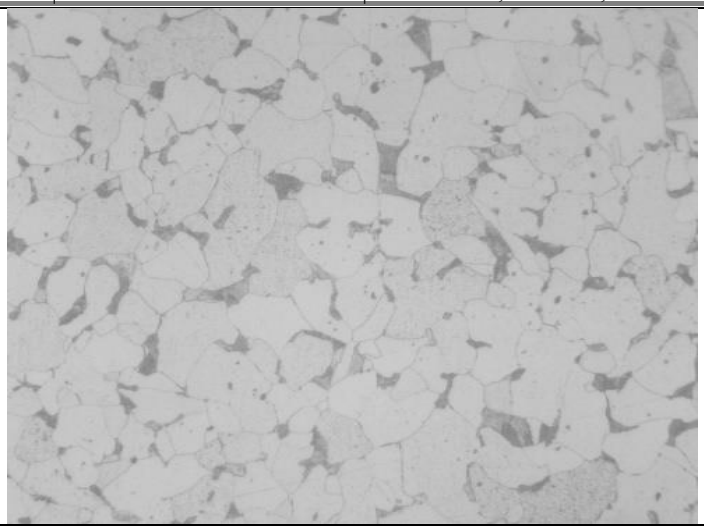
	
100X	500X

Figure 3: Hardening (Neem Bath)

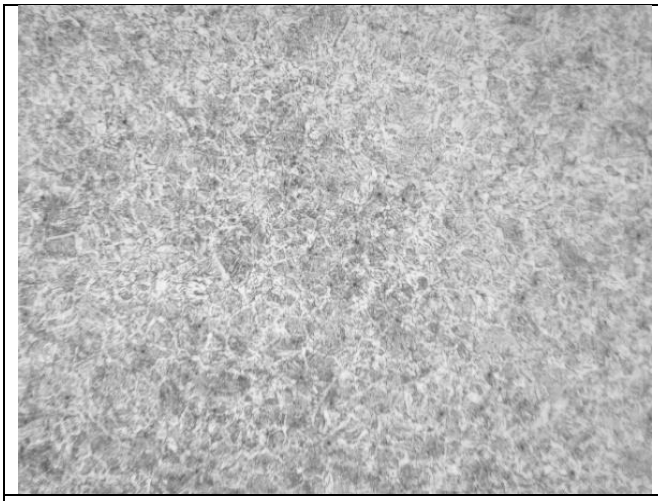
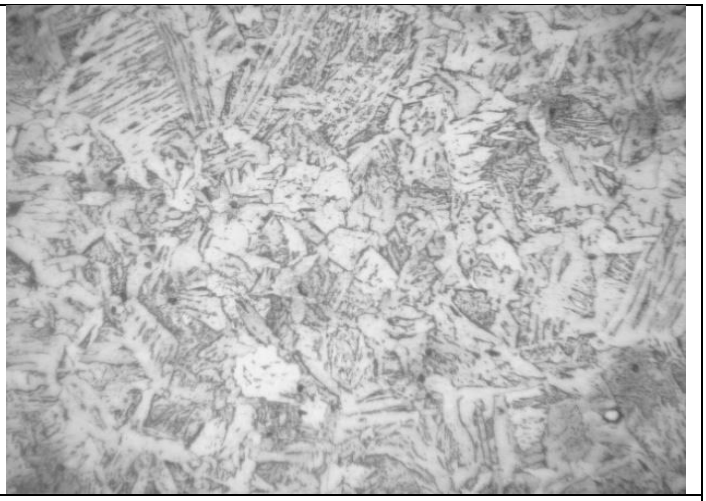
	
100X	500X

Figure 4: Hardening (Ice Bath)

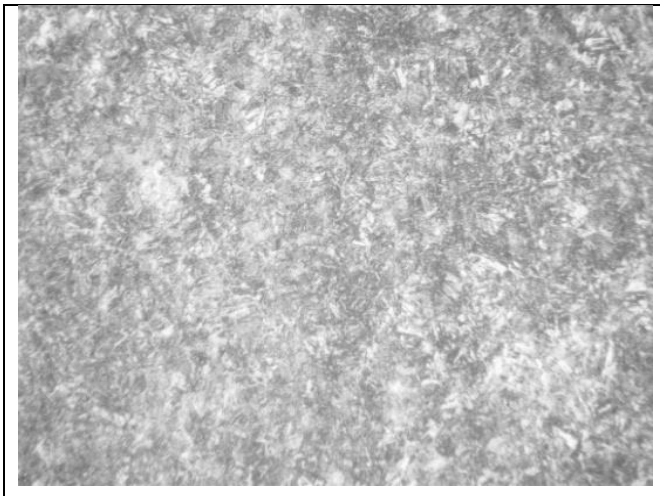
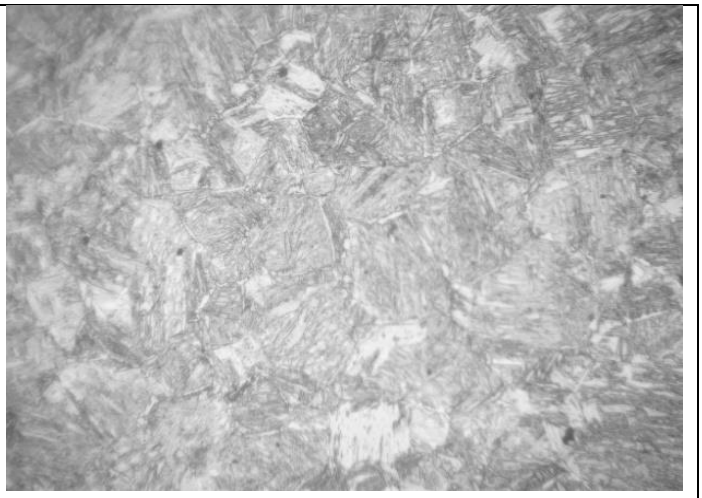
	
100X	500X

Figure 5: Hardening (Salt Bath)

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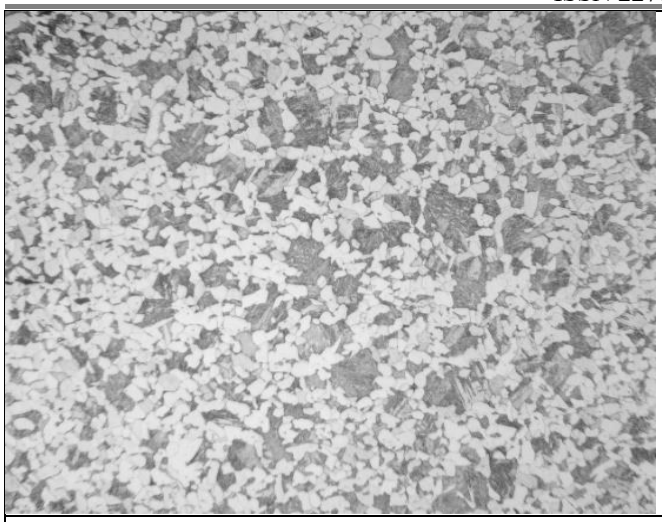
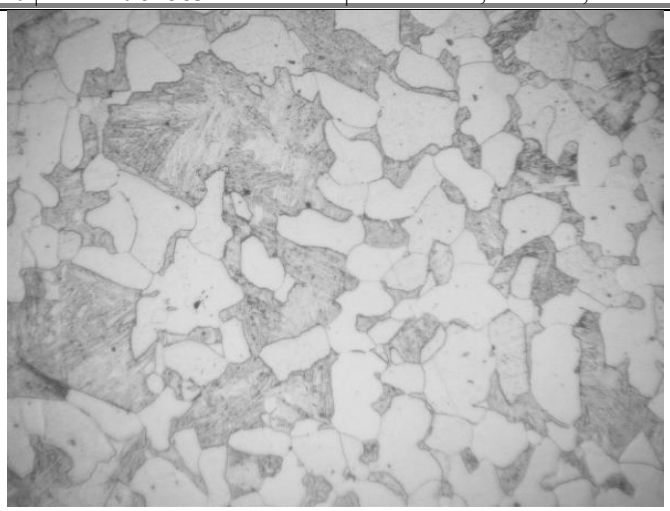
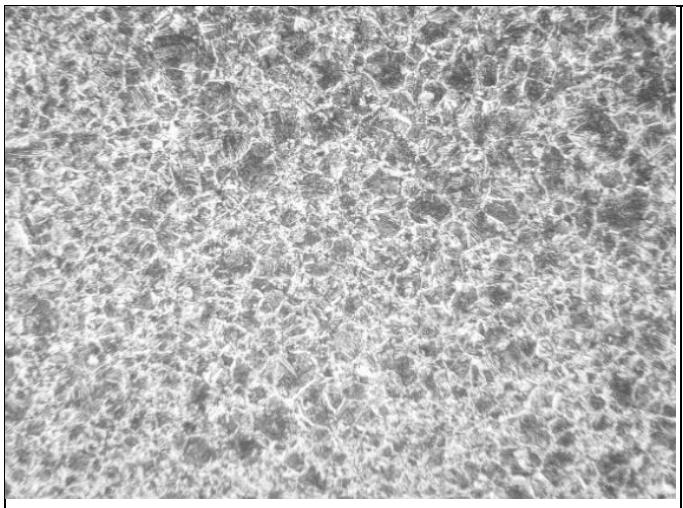
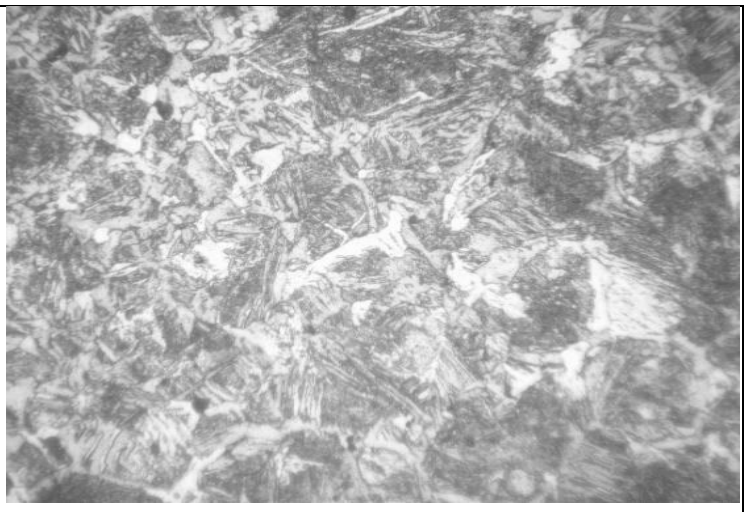
	
100X	500X

Figure 6: Hardening (Brine Solution)

	
100X	500X

RESULTS AND DISCUSSION

Table 2. Chemical analysis of low carbon steel by spectroscopy

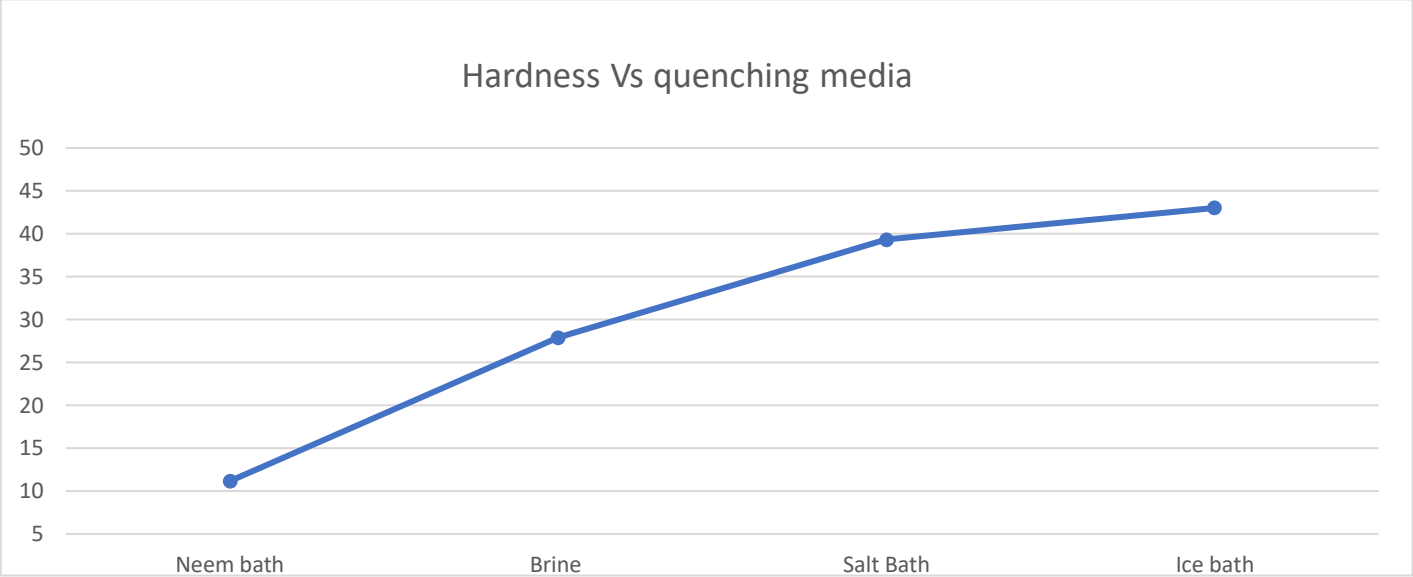
Element	Observed Value (%)
Carbon	0.213
Silicon	0.238
Manganese	0.627
Phosphorus	0.0275
Sulphur	0.0171

Table 3. Hardness value of heat Treatment Condition

Heat Treatment & Quenching Media	Average Hardness (HRC)
Furnace Cooling (Annealing)	11
Air Cooling (Normalizing)	13-14
Neem Bath	11.16
Ice Bath	43
Salt Bath	39.33
Brine Solution	27.91

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Graph 1: Hardness vs Quenching media



CONCLUSION

Neem bath quenching resulted in a refined ferrite–pearlite structure with fragmented cementite, indicating the initiation of a partially bainitic structure.

Annealing and normalizing produced a uniform distribution of ferrite and pearlite, leading to a stable and homogeneous microstructure.

Grai size Generally in annealing is 4 -5 and Normalizing Grain Size 6-8.

Quenching in brine solution and ice bath led to the formation of fine martensitic laths, contributing to higher hardness.

Salt bath quenching resulted in a mixed microstructure of martensite and retained austenite.

The highest hardness was observed in ice bath quenched samples, while the lowest hardness was obtained in annealed and neem bath.

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